



**ISSN: 2454-9940**



**INTERNATIONAL JOURNAL OF APPLIED  
SCIENCE ENGINEERING AND MANAGEMENT**

**E-Mail :**  
**editor.ijasem@gmail.com**  
**editor@ijasem.org**

**[www.ijasem.org](http://www.ijasem.org)**

## An analysis of gel permeation chromatography and its use in pharmaceuticals

DR. G. VAMSEE KRISHNA,A.RAMYA SREE,N. VENKATESWARA RAO

---

**ABSTRACT:** Gel permeation chromatography (GPC), a potent analytical method, is based on the elution of dissolved molecules from a column packed with a porous gel in order to separate them by size. It is also a kind of molecular sieving chromatography in which materials are broken down into their component parts after being dissolved in a solvent and then passed through a porous column packing. Gel permeation chromatography is often employed for a wide range of biochemical applications, as well as for the chemical study of large molecule species (polymers). The most used technique for determining the molecular weight distributions of complicated polymers is gel permeation chromatography. This article offers a perspective on the use of gel permeation chromatography in the pharmaceutical industry.

---

**KEY WORDS:** Gel permeation chromatography, analytical method, porous gel, sieving chromatography.

---

### I. INTRODUCTION:

Chromatography is a division technique used in substance analysis. Chromatography is used to measure the size of each component and isolate combinations in a single step. Additionally, because of their overall scopes in this manner, it is currently recognised as arguably the most spectacular and flexible scientific approach available. There are many other chromatography strategies that are used, but two of them are typically used: gas chromatography and fluid chromatography. [1] Fluid chromatography includes gel penetration chromatography. Lathe and Ruthven developed SEC as a methodology for the first time in 1955. [2] J. C. Moore invented the gel saturation chromatography. [3] The concept of gel saturation Chromatography dates back to J.C. Moore of the Dow Chemical Company who examined the procedure in 1964 and who authorised Waters to develop the unique segment. Corporation, who in 1964 promoted this invention as a result.

Partitioning techniques include gel penetration

chromatography (GPC), also known as gel filtration chromatography (GFC) or strainer size rejection chromatography. Another subset of high-performance fluid chromatography isolates polymer atoms based on their hydrodynamic volumes. The detachment of proteins, polysaccharides, chemicals, and synthetic polymers is accomplished with this technique. It is widely used to determine how high sub-atomic weight polymers' atomic weights are distributed. In the process' early development, cross-connected polydextran beads with variable pore sizes (sephadex, Pharmacia, Sweden) were used as the fixed stage. However, in recent years, gel saturation chromatography (GPC) has emerged as arguably the most important technique for partitioning. Pollock et al. used GPC for the first time to guarantee the chain length of fructans. In essence, gel penetration chromatography is a form of fluid chromatography in which the solute atoms are deliberately hindered from permeating the section pressing's dissolvable-filled pores.

---

Department of PHARMACEUTICAL ANALYSIS<sup>1,2,3</sup>

NRI College Of Pharmacy,

Pothavarappadu Village, AgiripalliMandal, Krishna Dist, Andhra Pradesh PinCode:521212

---

## II. METHOD OF WORKING:

A size-dependent partition, also known as size-avoidance chromatography. Based on the hydrodynamic volume or size of the analytes, GPC isolates. [4] This is distinct from other division methods, which segregate analytes based on substance or real associations. Gel has been poured into the portion. Gel functions as a stage that is fixed. Permeable dabs are used to create gel. Polystyrene, dextran, polyacrylamide, and agarose gels were all used as gels; they all had permeable designs.

the exact volume of gel involved in the segment. Maximum volume  $V_t = V_g$  (volume of the gel dabs) plus  $V_i$  (inward volume) plus  $V_o$  (free volume not

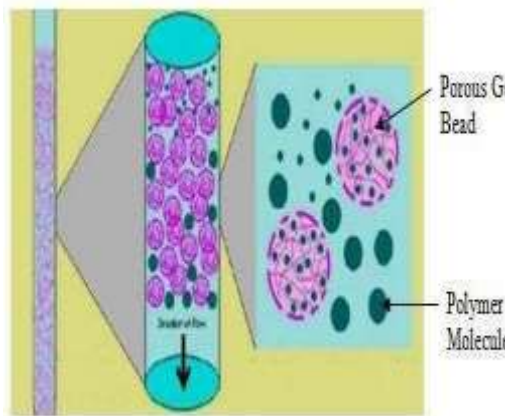


Figure 1: gel permeation chromatography

## III. INSTRUMENTATION OF GEL PERMEATION CHROMATOGRAPHY:

### 3a. Instrumentation:

Gel permeation chromatography is almost conducted in chromatography columns. The experimental design of gel permeation chromatography is slightly different from other techniques of liquid chromatography. Samples are dissolved in a proper solvent, in the case of GPC these tend to be organic solvents and after filtering the solution it is inserted onto a column. In the column the separation of multi-component mixture takes

occupied by the particles). Analyses  $V_t = V_g + V_i + V_o$

that are not held are eluted with the free volume outside of the particles ( $V_o$ ), while analytes that are totally held are eluted with volume of dissolvable held in the pores ( $V_i$ ). The all out volume can be considered by the accompanying condition, where  $V_g$  is the volume of the polymer gel and  $V_t$  is the complete volume. In the event that the example particles are little, they can without much of a stretch enter the pores of the globules and if test particles are huge, they neglect to enter the pores of the dabs.

The highest point of this section is appended with siphon which ceaselessly siphon portable stage in the segment. Base is associated with the indicator, the identifier the finder might be

1. Refractive list 2. UV retention 3. IR ingestion. Identifier is chosen dependent on the example.

At the point when the example alongside portable eliminator passed this section enormous particles outside the dab effectively push ahead where little atoms which are caught in the pores of the dabs set aside long effort to move [5].

place. With the help of a pump, the constant supply of fresh eluent to the column is accomplished. Detector is used because most analytes are not visible to the naked eye. To gain additional information about the polymer sample multiple detectors are used [6]. The availability of a detector makes the fractionation accurate and convenient.



**Figure 2: Instrument of gel permeation chromatography**

### 3b. Gel/stationary phase:

A gel is a semi-solid substance that can have properties ranging from soft and weak to hard and tough. [7] Gels are defined as a substantially dry cross-linked system, which exhibits no flow when in the steady-state. [8] A gel has been defined phenomenologically as a soft, solid or solid-like material consisting of two or more components, one is liquid, present in substantial quantity. In GPC gel is used as a stationary phase. In order to apply the gel to a given separation, the pore size of a gel must be carefully controlled. Other desirable properties of the gel for forming a gel are low affinity for the substance to be separated, and absence of ionizing groups. [9][10] Generally PL gel and Styragel (crosslinked), LH-20 (hydroxylpropylated sephadex), BioGel (crosslinked polyacrylamide), HW-20 and [12] HW-40 (hydroxylated methacrylic polymer), [13] agarose gels are often used in separation.

### 3c. Column:

In GPC column is filled with microporous packing material. The column is filled with gel. Inside the column separation of sample takes place, a hollow tube tightly packed with extremely small porous beads, polymer or silica have well defined pores size. Primarily for different molecular weight ranges, columns are packed with different sized particles with different sized pores. To improve the resolution, columns are usually employed in combination of two or three columns. Before the main line guard columns are used.

The guard column protects the main column by stopping insoluble particles or contaminants that could block the main column set.

Any of the following kinds may be

used: Analytical column-7.5-8mm diameters. Preparative columns-22-25mm Usual column length-25, 30, 50, and 60cm. Narrow-bore columns-2-3mm diameter have been introduced.

### 3d. Eluent/Mobile phase:

The eluent should be permitted high detector response from the polymer, should be a good solvent for the polymer and should wet the packing surface. The common eluent for polymers that disperse the remainder of the system [11].

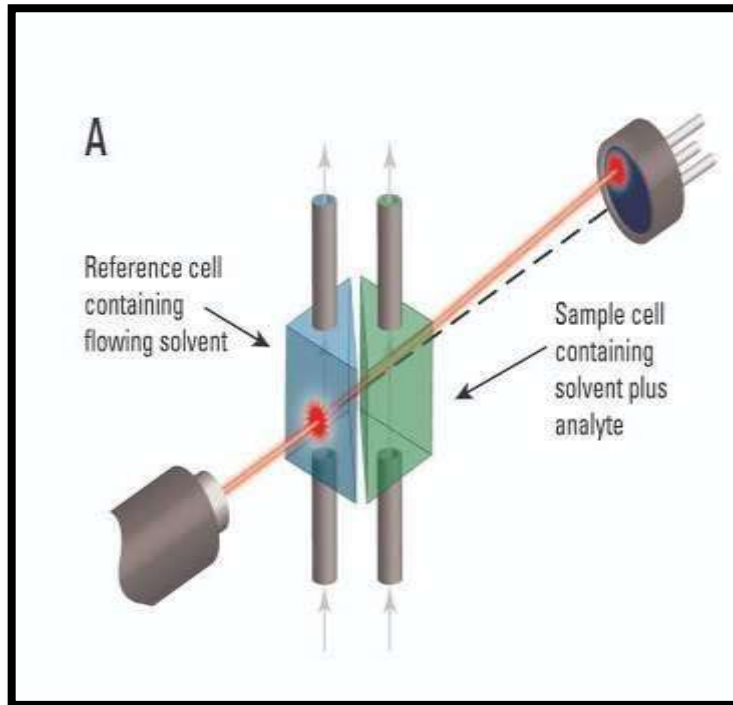
### 3e. Pump:

At constant, accurate and reproducible flow, the pump takes the solvent and delivers it to the remainder of the system. The pump has to be ready to run an equivalent flow regardless of viscosity, in order that results are often compared from one analysis to another. For uniform delivery of relatively small liquid volumes there are two sorts of pumps available they are: piston and peristaltic pumps. The pressure delivered by the pump to be smooth in order that there are not any pulses within the flow. The solvent is not wasted, when the inner volume of the pump is little. Pumps are very expensive in the equipment because they need to be made of chrome steel, titanium [12] and ceramics, which do not react with the solvents utilized in GPC. They ought to withstand very high pressures. For uniform delivery of relatively small liquid volumes there are two sorts of pumps available they are: piston and peristaltic pumps.

### 3f. Detectors:

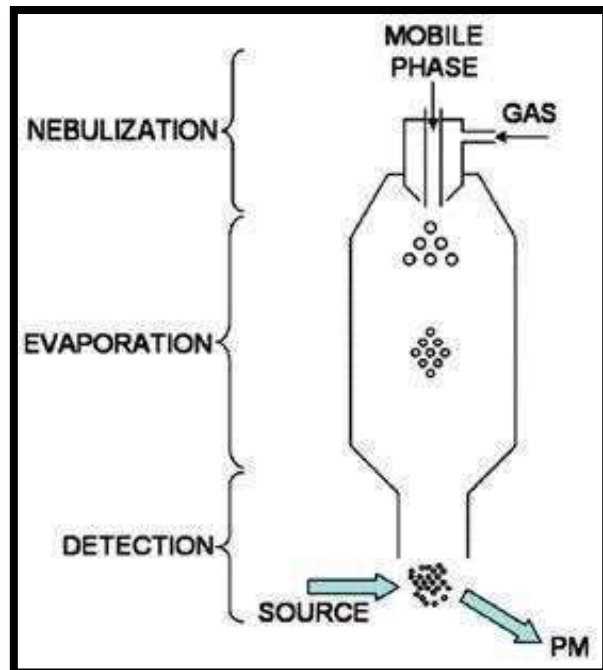
To detect their presence as they elute from a column, chromatography uses the chemical and physical properties of sample molecules and mobile phase and so different detectors have been developed that make use of the different character

istics of compounds. Detectors can be divided based on measure concentration alone, such as UV, differential refractive index (DRI), and evaporative light scattering (ELS) detectors, and those whose response is proportional to concentration and other properties of the polymer molecules, such as viscometers or static light scattering detectors. The most common gel permeation chromatography [13] detector is based on the principle of refractive index. In GPC, the concentration by weight of polymer in the eluting solvent may be monitored continuously with a detector. The first is concentration sensitive detectors which includes UV absorption, differential refractometer (DRI) or refractive index (RI) detectors, infrared (IR) absorption and density detectors. [14] The second category is molecular weight sensitive detectors, which include low angle light scattering detectors (LALLS) and multi angle light scattering (MALLS). The determination of most copolymer composition is done using UV and RI detectors, although other combinations can be used.



**Figure3.Differentialrefractiveindex:**

This detector estimates the difference in refractive index between the solvent and the eluting polymer solution. It can be used with almost any polymer solvent combination.



**Figure4:Evaporativelightscattering:**

It is the universal detector and functions by nebulizing column effluent into droplets, which are then evaporated in a heated gas stream and solvent remains in vapour. It is scattered by Mie scattering phenomenon and this intensity of scattered light is detected by a photomultiplier tube. The carrier gas flow causes nebulization of solutes and temperature causes evaporation in drift tube to form small particles of non-volatile solute.

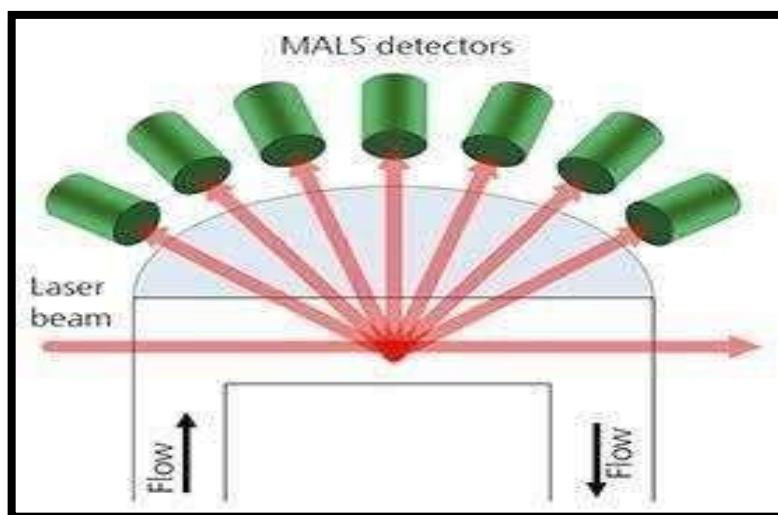


figure5: Multiangle light scattering:

The light scattering is an inherent limitation in determining the size of very small polymer molecules in solution. Molecular weight of a small polymer can be estimated by measuring the scattered-light intensity at 90° to the incident angle.

#### Steps in

**Gel Permeation Chromatography:** It involves three major steps [15]

##### A. Preparation of column for gel filtration

It involves:

1. Swelling of the gel
2. Packing the column semi-permeable, porous polymer gel beads with a well-defined range of pore sizes.
3. Washing: After packing, several column volumes of buffer solution is passed through the column to remove any air bubbles and to test the column homogeneity.

##### B. Loading the sample onto the column using a syringe

##### C. Eluting the sample and detection of components

#### IV. APPLICATIONS:

1. Gel permeation chromatography truly measures molecular volume and shape. GPC is used to determine

the relative molecular weight of polymer samples.  
2. It also determines the distribution of molecular weights.

##### Biochemical applications:

SEC can determine the quaternary structure of purified proteins. It can also be used for measurement of hydrodynamic volume with folded and unfolded versions of the same proteins. Eg: hydrodynamic radius of a typical protein domain can be 14 Å and 36 Å for the folded and unfolded forms respectively.

##### Surfactants study:

Non-ionic surfactants contain water-soluble group, and fatty acids. Various synthetic polymers contain hydrophobic segments. g.: Tween, Igepal, Brij, Pluronic, Triton etc. the analysis of surfactants is possible using SEC analysis and differential refractive index detector.

##### Polyesters analysis

Aromatic polyester analysis requires high temperature solvent due to its crystalline structure. It uses viscous m-cresol or o-chlorophenol as the mobile phase which causes hydrolytic degradation of the polymer [16]

##### Polycarbonates property study

Polycarbonates are non-crystalline thermoplastic and are linear aromatic polyesters of carbonic acids. The properties of polycarbonate like strength, clarity, and heat deflection temperatures. Can be studied by SEC.

#### **Polyamides property study:**

Polyamides are used as adsorbent for isolation of various polyphenol compounds from plants and as packaging material for different pharmaceutical products. SEC is employed during determinations of molecular weight distribution of polyamides to study different key physical parameters like strength, toughness, abrasion resistance, and retention of physical and mechanical properties.

#### **Natural rubber property study**

The natural rubber is widely used as pharmaceutical aid. Eg. new drug delivery system, rubber latex, contraceptive packaging material, etc. Hence the quality control check of natural rubber is essential to ensure its suitability for different applications. SEC is used to distinguish natural rubber from other conventional polymers.

5. Gel permeation chromatography can be used on humic acids or fulvic acids and in water and is usually applied to the analysis of fatty samples like fish.

6. It determines the quaternary structure of purified proteins.

#### **V. ADVANTAGES:**

It has many advantages. It has a well-defined separation time. Most samples can be analysed in an hour or less. It provides narrow bands. There is a lower chance for analyte loss to occur, since the analytes do not interact chemically or physically with the column. Molecular weights and mass distributions typically were not analysed, as these processes were quite labor-intensive. It is quick and relatively easy estimation of molecular weights and distribution for polymer samples.

#### **VI. DISADVANTAGES:**

Filtrations must be performed before using the instrument to prevent dust and other particulates from ruining the columns and interfering with the detectors. Although useful for protecting the instrument, there is the possibility of pre-filtration of the sample removing higher molecular weight sample before it can be loaded on the column. Another possibility to overcome these issues is the separation by field-

flow fractionation (FFF). Another disadvantage of GPC for polymers is that filtrations must be performed before using the instrument to prevent dust and other particulates from ruining the columns and interfering with the detectors.

#### **VII. CONCLUSION:**

Gel permeation chromatography is a powerful analytical technique. The preference for GPC is because of its relatively low cost, simplicity, and ability to provide accurate, reliable information about the molecular weight distribution of a polymer. It is used to determine the relative molecular weight of a polymer. Most samples can be analysed in an hour.

#### **REFERENCES:**

- [1]. Lathe, G.H.; Ruthven, C.R. *J. 1956, The Separation of Substance and*, 62, 665–674.
- [2]. H. Dai., P.L. Dubin, and, T. Andersson. 1998, *Permeation of Small Molecules in Aqueous Size Exclusion Chromatography Vis-à-Vis Models for Separation. Analytical Chemistry*, 70(8), 1576.
- [3]. Moore, J.C. 1964, *Gel permeation chromatography. I. A new method for molecular weight distribution of high polymers. J. Polym. Sci.*, 2, 835-843.
- [4]. Skoog, D.A. 2006, *Principles of Instrumental Analysis*, 6th ed.; Thompson Brooks/Cole: Belmont, California, Chapter 28.
- [5]. Khademhosseini A, Demirci U (2016). *Gels Handbook: Fundamentals, Properties and Applications*. World Scientific Pub Co Inc.
- [6]. Seiffert S, ed. (2015). *Supramolecular Polymer Networks and Gels*. Springer. ASIN B00VR5CMW6
- [7]. Ferry JD (1980). *Viscoelastic Properties of Polymers*. New York: Wiley.
- [8]. Almdal, K.; Dyre, J.; Hvidt, S.; Kramer, O. (1993). "Towards a phenomenological definition of the term 'gel'". *Polymer Gels and Networks*.
- [9]. Agilent Technologies. "agilent organic gpc/seccolumns" 2019-12-06.
- [10]. Waters corporation. "styragel column care and use manual" 2019-12-06.
- [11]. Gehealthcare. "sephadexlh (<https://cdn.gelifesciences.com/dmm3bwsv3/assetstrea>)
- [12]. M. a spx?mediaformatid=10061&destinationid=10016&assetid=11273. 2019-12-06.



- [13].12.tosohbioscience."toyoparlhw retrieved 2019-12-06.
- [14].A Laboratory Handbook for Gel Chromatography, Gel Filtration, Gel Permeation, and Molecular Sieves was published in 1969 by Helmut, D. Springer-Verlag,
- [15].Determining MWD and Chemical Composition of Polymers by Chromatographic Methods, B. Trathnigg. 1995, 20, 615–650, Prog.Polym. Sci.[2]
- [16].Adv. Polym. Sci. 2000, 150, 1-66. Pasch, H. Hyphenated Techniques in Liquid Chromatography of Polymers.[3]